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Efficient, Regioselective Access to Bicyclic Imidazo[1,2-x]- Heterocycles via Gold- and **Base-Promoted Cyclization of 1-Alkynylimidazoles**

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Reactions of 1-alkynylimidazoles involving the formation of their 2-lithio derivatives followed by addition of aldehydes or ketones are presented. The method gives access to 1-alkynyl-2-(hydroxymethyl)imidazoles which undergo 6-endo-dig or 5-exo-dig cyclization under AuCl₃or base-catalyzed conditions to yield imidazo[1,2-c]oxazoles and imidazo[2,1-c][1,4]oxazine heterocycles. Under transition metal catalysis, the reaction occurs in a regiospecific manner, leading exclusively to the product of 6-endo-dig attack, whereas under basic conditions, the reaction takes place in a regioselective manner giving preferentially the product from 5-exo-dig attack.

Imidazoles are an extremely important class of compounds with applications across all areas of chemistry,¹ biology,² and material science.³ In particular, fused bicyclic imidazoles, both naturally occurring and synthetic, display a wide range of biological activities⁴ and are found in a number of clinically important drugs.5 There has been a growing interest in methods to efficiently access known fused imidazole frameworks⁶ and to explore previously unreported

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chemotypes.⁷ Yet, many fused bicyclic imidazole systems remain un- or under-explored.6b

One promising approach to the rapid elaboration of a wide variety of fused bicyclic imidazo[1.2-x] heterocycles, having a bridgehead imidazole nitrogen atom, focuses on the cyclization reactions of the relatively unexplored 1-alkynylimidazoles.⁸ We recently reported a coupling reaction between imidazoles and bromoalkynes mediated by copper complexes allowing access to a wide range of these compounds.⁹ The further functionalization of those 1-alkynylimidazoles at the 2 position by formation of their 2-lithio derivatives and subsequent treatment with various electrophiles has been studied.¹⁰ The cyclization of 1-alkynylimidazoles containing appropriate nucleophilic groups at their 2 position can afford two types of bicyclic imidazoles depending upon the regiochemistry of the cyclization (Figure 1). Of particular interest is the degree to which the ynamine-like chemistry of the imidazole-substituted alkyne group plays a role in biasing the regiochemistry of this cyclization as well as the potential to control the regiochemistry by reagent selection using transition metals to activate the alkyne π -system.¹¹

In exploring the 2-functionalization of the 1-alkynylimidazole 1a by deprotonation and trapping of the anion with

(5) (a) Harrison, T. S.; Keating, G. M. CNS Drugs 2005, 19, 65–89. (b) Basiuk, V. A. Russ. Chem. Rev. 1997, 66, 187–204. (d) Ansini, M.; Cappelli, Jastak, V. T. Kats, China K. J. Karger, T. Bruni, G.; Romeo, R.; Basile, A. S. J. Med. Chem. 1996, 39, 4275–4284. (c) Heeres, J.; Backx, L. J. J.; Mostmans, (6) (a) Preston, P. N. Condensed Imidazoles: 5-5 Ring Systems; Wiley: New

York, 1986; pp 1 – 46. (b) Hulme, C.; Lee, Y.-S. *Mol. Diversity* **2008**, *12*, 1–15. (c) Nodwell, M.; Pereira, A.; Riffell, J. L.; Zimmerman, C.; Patrick, B. O.; Roberge, M.; Andersen, R. J. J. Org. Chem. **200**9, 77, 995–1006. (d) Allin, S. M.; Barton, W. R. S.; Bowman, W. R.; Bridge, E.; Elsegood, M. R. J.; McInally, T.; McKee, V. Tetrahedron 2008, 64, 7745–7758. (e) Lovely, C. J.; Du, H.; Sivappa, R.; Bhandari, M. R.; He, Y.; Dias, H. V. R. J. Org. Chem. 2007, 72, 3741–3749. (f) Chen, Y.; Dias, H. V. R.; Lovely, C. J. Tetrahedron Lett. 2003, 44, 1379–1382.

 (i) Ghu, L., Bias, H. V., Bovi, S. V. M. (2005), 47, 1971 [2005], 47, 1971 [2005].
 (g) Hua, D. H.; Zhang, Z.; Chen, J. J. Org. Chem. 1994, 59, 5084–5087.
 (7) (a) Krasavin, M.; Shkavrov, S.; Parchinsky, V.; Bukhryakov, K.
 J. Org. Chem. 2009, 74, 2627–2629. (b) Nowak, J.; Skalski, B.; Gdaniec, Z.; Milecki, J. Tetrahedron Lett. 2009, 50, 1671-1673. (c) Nayak, M.; Kanojiya, S.; Batra, S. Synthesis 2009, 431–437. (d) Gracias, V.; Gasiecki,
 A. F.; Djuric, S. W. Org. Lett. 2005, 7, 3183–3186. (e) Le Bas, M.-D. H.;
 O'Shea, D. F. J. Comb. Chem. 2005, 7, 947–951.

(10) Laroche, C.; Kerwin, S. M. Tetrahedron Lett. 2009, 50, 5194-5197.

^{(1) (}a) Grimmett, M. R. Imidazole and Benzimidazole Synthesis; Academic Press: New York, 1997; pp 1–143. (b) Bellina, F.; Cauteruccio, S.; Rossi, R. *Tetrahedron* **2007**, *63*, 4571–4624.

⁽²⁾ For review, see: De Luca, L. Curr. Med. Chem. 2006, 13, 1-23.

^{(3) (}a) Moylan, C. R.; Miller, R. D.; Twieg, R. J.; Betterton, K. M.; Lee, V. Y.; Matray, T. J.; Nguyen, C. *Chem. Mater.* **1993**, *5*, 1499–1508.
 (b) Arduengo, A. J.; Krafczyk, R.; Schmutzler, R.; Craig, H. A.; Goerlich, J. R.; Marshall, W. J.; Unverzagt, M. Tetrahedron 1999, 55, 14523-14534.

⁽⁴⁾ For recent examples, see: (a) Kim, P.; Kang, S.; Boshoff, H. I.; Jiricek, J.; Collins, M.; Singh, R.; Manjunatha, U. H.; Niyomrattanakit, P.; Zhang, L.; Goodwind, M.; Dick, T.; Keller, T. H.; Dowd, C. S.; Barry, C. E. J. Med. Chem. 2009, 52, 1329–1344. (b) Thompson, A. M.; Blasser, A.; Anderson, R. C.; Reilly, P.; Rothlein, R.; Sellati, R. H.; Woska, J. R., Jr.; Chen, S.; Gunn, J. A.; O'Brien, D.; Norris, S. H.; Kelly, T. A. J. Med. Chem. 2007, 47, 5356-5366. (d) Gehlert, D. R.; Cippitelli, A.; Thorsell, A.; Le, A.-D.; Hipskind, P. A.; Hamdouchi, C.; Lu, J.; Hembre, E. J.; Cramer, J.; Song, M.; McKinzie, D.; Morin, M.; Ciccocioppo, R.; Helig, M. J. Neurosci. **2007**, *27*, 2718–2726. (e) Enguehard-Gueiffier, C.; Gueiffier, A. Mini-Rev. Med. Chem. **2007**, 7, 88– 899. (Ĭ) Miwa, S.; Mizokami, A.; Keller, E. T.; Taichman, R.; Zhang, J.; Namiki, M. *Cancer Res.* **2005**, *65*, 8818–8825. (g) Dubuisson, M. L. N.; Ress, J.-F.; Marchand-Brynaert, J. *Drug Dev. Ind. Pharm.* **2005**, *31*, 827-849. (h) Chimirri, A.; Monforte, P.; Rao, A.; Zappala, M.; Monforte, A. M.; De Sarro, G.; Pannecouque, C.; Witvrouw, M.; Balzarini, J.; De Clercq, E. *Antiviral Chem. Chemother.* **2001**, *12*, 169–174.

⁽⁸⁾ For cyclizations of 1,2-dialkynylimidazoles to imidazo[1,2a]pyridines, see: Nadipuram, A.; Kerwin, S. M. Tetrahedron 2006, 62, 3798-3808.

⁽⁹⁾ Laroche, C.; Li, J.; Freyer, M. W.; Kerwin, S. M. J. Org. Chem. 2008, 73. 6462-6465



FIGURE 1. Regioselective one-pot conversion of 1a to 3a.

benzaldehyde, the formation of the 5,7-dihydroimidazo[1,2-c]oxazole **3a** was observed.^{10,12} The structure of **3a** was confirmed by X-ray crystallography.¹⁰ Interestingly, none of the 6-*endo-dig* cyclization product **4a** was present.¹⁴

A number of proposals have been put forth to rationalize the 5-exo-dig versus 6-endo-dig cyclization modes¹³ of 4ynols under basic conditions, including the double bond stereochemistry of the 5-exo cyclization products.^{12,13,15} Recent studies have demonstrated alternative means to affect the regioselectivity of these cyclizations through metal catalysis.¹⁶ However, it was not clear at the outset how applicable these precedents would be to the electronically biased 1-alkynylimidazole system **1a**. Thus, in order to

(12) For related 5-exo-dig cyclizations, see: (a) Chai, Z.; Xie, Z.-F.; Liu,
X.-Y.; Zhao, G.; Wang, J.-D. J. Org. Chem. 2008, 73, 2947–2950. (b)
Marshall, J. A.; DuBay, W. J. J. Org. Chem. 1994, 59, 1703–1708. (c)
Marshall, J. A.; Bennett, C. E. J. Org. Chem. 1994, 59, 6110–6113.

Marshall, J. A., Dubay, W. J. S. Org. Chem. 1994, 59, 6110–6113.
 (13) (a) Baldwin, J. E. J. Chem. Soc., Chem. Commun. 1976, 734–735. (b) Johnson, C. D. Acc. Chem. Res. 1993, 26, 476–482. (c) Sheng, Y.; Musaev, D. G.; Reddy, K. S.; McDonald, F. E.; Morokuma, K. J. Am. Chem. Soc. 2002, 124, 4149–4157.

(14) For reports of analogous, base-promoted 6-endo-dig cyclizations, see: (a) Hiroya, K.; Jouka, R.; Kameda, M.; Yasuhara, A.; Sakamoto, T. Tetrahedron **2001**, *57*, 9697–9710. (b) Brennan, C. M.; Johnson, C. D.; McDonnell, P. D. J. Chem. Soc., Perkin Trans. 2 **1989**, 957–961. (c) Garcia, H.; Iborra, S.; Primo, J. J. Org. Chem. **1986**, *51*, 4432–4436.

(15) (a) Nakatani, K.; Okamoto, A.; Saito, I. Tetrahedron 1996, 52, 9427–9446. (b) Houk, K. N.; Strozier, R. W.; Rozeboom, M. D.; Nagase, S. J. Am. Chem. Soc. 1982, 104, 323–325. (c) Wipf, P.; Graham, T. H. J. Org. Chem. 2003, 68, 8798–8807. (d) Ivanchikova, I. D.; Usubalieva, G. E.; Schastnev, P. V.; N., E.; Moroz, A. A.; Shvartsberg, M. S. Russ. Chem. Bull. Int. Ed. 1992, 41, 1672–1679. (e) Padwa, A.; Krumpe, K. E.; Weingarten, M. D. J. Org. Chem. 1995, 60, 5595–5603. (f) Nakatani, K.; Okamoto, A.; Yamanuki, M.; Saito, I. J. Org. Chem. 1994, 59, 4360–4361. (g) Evans, C. M.; Kirby, A. J. J. Chem. Soc., Perkin Trans. 2 1984, 1269–1275.

(16) (a) Ramana, C. V.; Mallik, R.; Gonnnade, R. G. Tetrahedron 2008, 64, 219–223. (b) Koo, B.; McDonald, F. E. Org. Lett. 2007, 9, 1737–1740. (c) Liu, Y H.; Song, F. J.; Song, Z. Q.; Liu, M.; Yan, B. Org. Lett. 2005, 7, 5409–5412. (d) Antoniotti, S.; Genin, E.; Michelet, V.; Genët, J.-P. J. Am. Chem. Soc. 2005, 127, 9976–9977. (e) Genin, E.; Antoniotti, S.; Michelet, V.; Genet, J.-P. Angew. Chem., Int. Ed. 2005, 44, 4949–4953. (f) Mzhelskaya, M. A.; Ivanchikova, I. D.; Polyakov, N. E.; Moroz, A. A.; Shvartsberg, M. S. Russ. Chem. Bull. Int. Ed. 2004, 53, 2798–2804. (g) Gabriele, B.; Salerno, G.; Fazio, A.; Pittelli, R. Tetrahedron 2003, 59, 6251–6259. (h) Qing, F.-L.; Gao, W.-Z. Tetrahedron Lett. 2000, 41, 7727–7730. (i) Gabriele, B.; Salerno, G.; Lauria, E. J. Org. Chem. 1999, 64, 7687–7692. (j) Dalla, V.; Pale, P. New J. Chem. 1999, 803–805. (k) Gabriele, B.; Salerno, G. Chem. Commun. 1997, 1083–1084. (l) Marshall, J. A.; Sehon, C. A. J. Org. Chem. 1995, 60, 5966–5968. (m) Seiller, B.; Bruneau, C.; Dixneuf, P. H. Tetrahedron 1995, 51, 13089–13102.

TABLE 1. Catalyst Screen for 6-endo-dig Cyclization



^aBased on recovered starting material. ^bIsolated yield after chromatography.

explore the generality of the cyclization to 3a and to develop a means of accessing the 6-*endo-dig* cyclization product 4a, the isolation and characterization of the carbinol derived from the anion 2a' were undertaken.

Compound **2a** (Table 1) can be isolated in 95% yield from the reaction of 1-alkynylimidazole **1a** under the conditions shown in Figure 1 when the reaction mixture is quenched with 1 M HCl solution rather than water. Treatment of alcohol **2a** in the presence of various metal catalysts in refluxing CH_3CN^{17} was investigated (Table 1). The reaction proceeded to completion using AuCl₃ as catalyst affording imidazo[2,1-*c*][1,4]oxazine **4a** (Table 1, entry 2) as the only product in 95% isolated yield.¹⁸

In addition to AuCl₃, AuCl (Table 1, entry 1), $PdCl_2$ (Table 1, entry 3), and CuCl₂ (Table 1, entry 6) afforded the cyclized compound **4a**, although in lower yield. Interestingly, none of these cases showed any traces of the 5-*exo-dig* product **3a**.

Our interest in the formation of both compounds **3a** and **4a** led us to reinvestigate the cyclization reaction of **2a** under basic conditions. Under aqueous conditions (1 equiv of 1 M NaOH) in THF at room temperature, the desired compound **3a** was obtained in 75% yield. The use of strong organic bases (*n*-BuLi or EtMgBr) in THF did not lead to any conversion at room temperature and led to degradation under forcing conditions. A catalytic quantity of K_3PO_4 (5 mol %) in CH₃CN at reflux afforded **3a** in 80% yield within an hour. The cyclization was also attempted under acidic conditions (HCl saturated THF) with no success.¹⁹ A few 5,7-dihydroimidazo[1,2-*c*]oxazole derivatives have been reported in the literature for their biological properties and are generally prepared by condensation reaction

^{(11) (}a) Patil, N. T.; Yamamoto, Y. Chem. Rev. 2008, 108, 3395–3442.
(b) Kirsch, S. F. Synthesis 2008, 3183–3204. (c) Hashmi, A. S. K. Chem. Rev. 2007, 107, 3180–3211. (d) Alvarez-Corral, M.; Munoz-Dorado, M.; Rodriguez-Garcia, I. Chem. Rev. 2008, 108, 3174–3198. (e) Gorin, D. J.; Toste, F. D. Nature 2007, 446, 395–403. (f) Fürstner, A.; Davies, P. W. Angew. Chem., Int. Ed. 2007, 46, 3410–3449. (g) Nakamura, I.; Yamamoto, Y. Chem. Rev. 2004, 104, 2127–2198. (h) Zeni, G.; Larock, R. C. Chem. Rev. 2004, 104, 2285–2309.

⁽¹⁷⁾ Other solvents screened include CH_2Cl_2 (5% conversion with 2 mol % of AuCl₃ after 14 h) and THF (100% conversion and 60% yield of **4a** with 2 mol % of AuCl₃ after 14 h, with the formation of several unidentified side products).

⁽¹⁸⁾ The imidazo[2,1-*c*][1,4]oxazine system has been reported previously only once: (a) Essassi, E. M.; Fifani, J.; Hamamsi, I. *Bull. Soc. Chim. Belg.* **1994**, *103*, 83–84. The related 6,8-dihydroimidazo[2,1-*c*][1,4]oxazines are also largely unexplored: (b) O'Sullivan, D. G.; Pantic, D.; Wallis, A. K. *Nature* **1965**, *205*, 262–264. (c) Medaer, B. P.; Hoornaert, G. J. *Tetrahedron* **1999**, *55*, 3987–4002.

⁽¹⁹⁾ This led to a rapid conversion of **2a** into compounds arising from the addition of HCl across the carbon–carbon triple bond, as has been previously observed for 1-alkynylimidazoles; see refs 8 and 9.

TABLE 2. Regioselective Cyclizations of 1-Alkynylimidazoles 2a-j



entry	R ₁	R ₂	R ₃	R_4	yield ^{<i>a</i>} (1 to 2)	yield ^{<i>a</i>} (2 to 3)	yield ^{<i>a</i>} (2 to 4)
1	Н	Ph	Н	Ph	2a (95%)	3a (80%)	4a (95%)
2	Н	Ph	Н	Et	2b (90%)	3b (97%)	4b (99%)
3	Н	Ph	Ph	Et	2c (86%)	3c (99%)	4c (100%)
4	Benz^b	Ph	Ph	Н	2d (87%)	3d (96%)	4d (85%)
5	4-Ph-im ^c	Ph	Ph	Н	2e (92%)	3e (78%)	4e (95%)
6	Н	$p-^{t}$ Bu-Ph	Н	Ph	2f (92%)	3f (93%)	4f $(100\%)^d$
7	Н	p-CF ₃ -Ph	Н	Ph	2g (85%)	3g (85%)	$4g(85\%)^{e}$
8	Н	TIPS	Н	Ph	2h (90%)	3h (86%)	4h (0%)
9	Н	Н	Н	Ph	2i ^f	3i (24%)	4i (16%)
10	Н	<i>n</i> -Hex	Н	Ph	2j (87%)	3 j/ 4 j (93%) ^g	4j (87%)
^a Isolate	d vield after chromat	ography. ^b Benz refers	s to the benzim	idazole core. ^{<i>c</i>}	4-Ph-im refers to the (4-Ph	n)-imidazole core. ^d Yield a	fter 1 h. ^e Yield after

6 days. ^fYield from **2h** (TBAF, THF, $-78 \,^{\circ}$ C) is 80%. ^g**3j** and **4j** were obtained as a 53/47 mixture but can be separated by flash chromatography.



FIGURE 2. Iodo-cyclization of 2a into 5a.

between 2-(hydroxymethyl)imidazole and aldehydes.²⁰ However, 5-methylene-5,7-dihydroimidazo[1,2-c]oxazole derivatives such as **3a**, which possess a cyclic *N*,*O*-ketene acetal poised for further elaboration,²¹ have not been previously reported.

In the same vein, the iodo-cyclization reaction of **2a** using excess of NIS and K_3PO_4 affords the iodo-imidazo[2,1-*c*]-[1,4]oxazine **5a** (Figure 2), which provides a chemical handle for further functionalization of the oxazine ring.^{22,23}

Having optimized the conditions for the cyclization of **2a** in either the 5-*exo-dig* or the 6-*endo-dig* mode, the scope of

(22) The structure of **5a** was confirmed by X-ray diffraction. **5a** was converted into **4a** by treating with magnesium followed by H_2O quench.

(23) Notably, Liu and co-workers reported compounds arising from a 5exo-dig process under similar conditions: Liu, Y.; Song, F.; Cong, L. J. Org. Chem. 2005, 70, 6999–7002. these transformations was evaluated. Toward this end, compounds 2b-j were synthesized, and their cyclizations were carried out in presence of AuCl₃ or K₃PO₄ in CH₃CN (Table 2).

Variation of the substituents at the carbinol center $(R_3,$ R_4) and at the imidazole core (R_1) offered alcohols 2a-e as well as the corresponding 5-exo-dig 3a-e and 6-endo-dig 4a-e cyclization products from good to excellent yields (Table 2, entries 1-5). Substrates possessing either an electron-donating (2f) or an electron-withdrawing (2g) aryl substituent on the carbon-carbon triple bond both led to the expected products in high yields (Table 2, entries 6 and 7); however, the rate of the AuCl3-catalyzed cyclization was dramatically affected. The cyclization of the 4-methoxy substituent analogue 2f in the presence of AuCl₃ is complete in just 1 h, whereas the cyclization of the (4-trifluoro)phenyl analogue 2g requires 6 days before reaching completion. The cyclization of the triisopropylsilyl-substituted 1-alkynylimidazole **2h** occurred in the presence of K_3PO_4 to give **3h** in 86% yield; however, no conversion of 2h into 4h in the presence of AuCl₃ was observed even after extended reaction times (Table 2, entry 8). Apparently, the steric hindrance of the bulky TIPS substituent prevents approach of the hydroxyl group to the terminal sp carbon required for the 6-endodig cyclization. In the case of the 1-ethynylimidazole 2i, prepared from **2h** by protodesilylation (TBAF, THF, -78 °C, 30 min), the corresponding 5-exo-dig 3i and 6-endo-dig 4i products were obtained in low yields (Table 2, entry 9). In these cases, the formation of multiple side products was observed. Finally, reaction of 2j under basic condition is the sole example where a mixture of compounds has been obtained (Table 2, entry 10). Under gold catalysis, 2j gave exclusively compound 4j from 6-endo-dig attack. Remarkably, with the exceptions of compound 2h, which failed to

^{(20) (}a) Katritzky, A. R.; Aslan, D. C.; Leeming, P.; Stell, P. J. *Tetrahedron: Asymmetry* **1997**, *8*, 1491–1500. (b) Katritzky, A. R.; Aslan, D. C.; Oniciu, D. C. *Tetrahedron: Asymmetry* **1998**, *9*, 2245–2251. (c) Chimirri, A.; Monforte, P.; Zappalà, M.; Monforte, A. M.; De Sarro, G.; Pannecouque, C.; Witvrouw, M.; Balzarini, J.; De Clercq, E. *Antiviral Chem. Chemother.* **2001**, *12*, 169–174.

⁽²¹⁾ For recent studies of cyclic *N*,*O*-ketene acetals, see: Zhou, A.; Cao, L.; Li, H.; Liu, Z.; Pittman, C. U., Jr. *Synlett* **2006**, 201–206. (b) Zhou, A.; Njogu, M. N.; Pittman, C. U., Jr. *Tetrahedron* **2006**, *62*, 4093–4102. (c) Huang, Y.-t.; Moeller, K. D. *Tetrahedron* **2006**, *62*, 6536–6550. (d) Quast, H.; Ach, M.; Balthasar, J.; Hergenroether, T.; Regnat, D.; Lehmann, J.; Banert, K. *Helv. Chim. Acta* **2005**, *88*, 1589–1609.

afford the 6-*endo-dig* cyclization product under AuCl₃ catalysis, and **2j**, which gave a mixture of **3j** and **4j** under basic condition, others alcohols **2** examined afforded exclusive 5-*exo-dig* cyclization in the presence of K_3PO_4 and exclusive 6-*endo-dig* cyclization in the presence of AuCl₃. We are unaware of a similar case of such high reagent control of both 5-*exo-dig* and 6-*endo-dig* cyclization modes in other 4-ynol cyclizations.

We believe that cycloisomerizations under basic and π -acid metal conditions are two distinct cases. Under basic condition, competition should take place between 5-exo-dig and 6-endo-dig cycloisomerization.^{15e} With 1-alkynylimidazole substrates, the nitrogen attached on the carbon-carbon triple bond should strongly disfavor the formation of the carbionic species at its α -position (6-endo-dig attack). Under transition metal catalysis, a complete regioselective process regardless of the metal or the electronic nature of the substituents is observed. This indicates that, under transition metal catalysis, the 5-exo-dig attack is strongly disfavored.^{13c} We propose that Au catalyzes the cycloisomerization by activation of the alkyne, for example, through a [$(\eta^2$ -alkyne)-AuX_n] complex as recently described by Behrens.²⁴ This coordination results in preferential acceleration of the 6-endo-dig cyclization, possibly due to the proximity of the nucleophile to the alkyne terminal carbon in this proposed gold η^2 complex.

In conclusion, we have shown that 1-alkynyl-2-(hydroxymethyl)imidazoles can be cyclized in the presence of K_3PO_4 or AuCl₃ in CH₃CN to afford preferentially 5-endodig or exclusively 6-endo-dig cyclization products, respectively. These cyclizations exemplify a powerful new strategy for the construction of various imidazo-fused heterocycles from 1-alkynylimidazoles.

Experimental Section

Typical Procedure for the Synthesis of Phenyl-(1-(2phenylethynyl)-1*H*-imidazol-2-yl)methanol (2a): To a solution of 1-(2-phenylethynyl)-1*H*-imidazole (1a) (168 mg, 1 mmol) in THF (10 mL) under argon at -78 °C was added *n*-BuLi (0.400 mL of 2.5 M solution in hexane, 1 mmol). The reaction mixture was stirred for 15 min at -78 °C prior to the addition of benzaldehyde (0.112 mL, 1.1 mmol). After stirring for 30 min at -78 °C, the mixture was quenched with 1 M HCl solution (5 mL) and the temperature was allowed to rise to room temperature. The reaction mixture was then extracted with CH₂Cl₂ (3 × 50 mL), the combined organic layers were dried over Na₂SO₄, and the solvent was evaporated under reduced pressure. The residue was purified by flash chromatography (0-50% EtOAc/hexane) to afford 260 mg (95%) of **2a** as a white crystalline solid: mp 89.8–90.9 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.47–7.42 (2H, m), 7.39–7.28 (8H, m), 7.13 (1H, s), 7.00 (1H, s), 6.05 (1H, s), 4.39 (1H, s, OH); ¹³C NMR (100 MHz, CDCl₃) δ 152.7 (br s), 140.3, 131.6 (2C), 129.0, 128.5 (2C), 128.4 (2C), 128.1, 127.6, 127.1 (2C), 122.3 (br s), 120.9, 77.1, 73.5, 69.5; IR (KBr) 3127, 2259, 1432, 1248, 1179, 1150, 1054, 758, 695 cm⁻¹; MS (CI) 549 (2M + 1, 4%), 531 (2M – 17, 8%), 275 (M + 1, 100%), 257 (M – 17, 23%); HRMS calcd for C₁₈ H₁₅ N₂ O (M + H⁺) 275.1184, found 275.1181.

Typical Procedure for the Synthesis of (Z)-5-Benzylidene-7phenyl-5,7-dihydroimidazo[1,2-c]oxazole (3a):. To a solution of 2a (69 mg, 0.25 mmol) in CH₃CN (5 mL) was added K_3PO_4 (3 mg, 0.0125 mmol). The mixture was gently refluxed until completion. After chilling, the solvent was evaporated under reduced pressure. The residue was purified by flash chromatography (0-20% EtOAc/hexane) to afford 55 mg (80%) of **3a** as a white solid: mp 107.6–109.8 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.62-7.57 (2H, m), 7.52-7.46 (2H, m), 7.45-7.38 (3H, m), 7.36-7.30 (2H, m), 7.26-7.22 (2H, m), 7.20-7.14 (1H, m), 6.50 (1H, s), 5.50 (1H, s); ¹³C NMR (100 MHz, CDCl₃) δ 150.8, 144.1, 135.9, 135.1, 133.8, 129.4, 128.9 (2C), 128.5 (2C), 127.3 (2C), 126.6 (2C), 125.7, 110.3, 85.4, 80.1; IR (KBr) 3126, 1704, 1536, 1403, 1286, 1155, 1030, 996 cm⁻¹; MS (CI) 549 (2M + 1, 3%), 275 (M + 1, 100%), 257 (M - 101, 20%), 157 (M - 117, 40%); HRMS calcd for $C_{18}H_{15}N_2O(M + H^+)$ 275.1184, found 275.1186.

Typical Procedure for the Synthesis of 6,8-Diphenyl-8Himidazo[2,1-c][1,4]oxazine (4a):. To a solution of 2a (69 mg, 0.25 mmol) in CH₃CN (5 mL) was added AuCl₃ (0.5 mL of a 0.01 M solution in CH₃CN, 0.005 mmol). The mixture was gently refluxed until completion. After chilling, the solvent was evaporated under reduced pressure. The residue was purified by flash chromatography (0-30% EtOAc/hexane) to afford 65 mg (95%) of **4a** as a white solid: mp 115–117 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.63-7.58 (2H, m), 7.46-7.41 (2H, m), 7.41-7.32 (6H, m), 7.16 (1H, d, J = 1.4 Hz), 7.07 (1H, d, J =1.4 Hz), 7.01 (1H, s), 6.49 (1H, s); ¹³C NMR (100 MHz, CDCl₃) δ143.0, 139.6, 136.7, 131.9, 129.2, 128.9, 128.7, 128.5 (2C), 128.4 (2C), 127.1 (2C), 124.4 (2C), 115.6, 102.0, 76.1; IR (KBr) 3112, 1652, 1484, 1449, 1437, 1400, 1337, 1162, 1026, 749 cm⁻¹; MS (CI) 275 (M + 1, 100%); HRMS calcd for $C_{18}\,H_{15}\,N_2\,O~(M+$ H⁺) 275.1184, found 275.1185.

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Supporting Information Available: General experimental conditions, detailed experimental procedures, X-ray structure of **5a**, ¹H NMR and ¹³C NMR spectra for all new compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.

⁽²⁴⁾ Schulte, P.; Behrens, U. Chem. Commun. 1998, 1633.